EXHIBIT B

STATE OF CALIFORNIA

Killy- help !

State Water Resources Control Board DIVISION OF WATER RIGHTS

901 P Street, Sacramento P. O. Box 2000, Sacramento, CA 95810

APPLICATION to APPROPRIATE WATER

(For explanation of entries required, see booklet "How to File an Application to Appropriate Water in California")

	1. A	APPLICANT						<u> </u>		
	1	CALIF. DE	PT OF 1	ARKS	AND A	PECREH	7/011			
		(Name of Applicant)				1.0.44	(Talashara N			
								m, and 5 p,m,—in	may be reached iclude area code.	
		(Address)		(City o	or Town)	(State)			(Zit Cote	
		do hereby make application (propriate the fo	llowing describe	d waters of the St	ate of California	2,		
		SUBJECT TO VESTED RIGI	łTS				twell.	underf	(m ?)	
		OURCE The name of the source	at the point of div	ercion is	AVIOTA	CREEK	1 001 11			
	2	. The name of the source	at the point of the	cision is	(If unr	named, state natur	e of source and	that it IS unname	ed)	
		tributary to					·			
	Ь	. In a normal year does th	e stream dry up at	lany point dow	nstream from yo	ur project? YES	□ NO	🔀 . If Yes, d	uring what month.	
		is it usually dry?					54	np Ha	6170 th	ACT
		OINT of DIVERSION			20	v nan	120	TWO IT O	NUESTRUA	~
	a	. The point of diversion w	ill be in the Cour	ity of AM	TA PAK	BARA	125	NOLA !	Der Te	KEFA
	ь	List all points giving	coordinate distan	ices from section	on corner	Point is with	in Sect	tion Township	Base and	
		or other tie	as allowed by Boa	ard regulations		(40-acre Subdivi	sion)		Meridan	- >
rsion		The deli				¥ of	y de		232 W	SBN
17 - 13 1	11.					¼ of	4			
ck w/ B11	19					1/4 of	y,			
		c. Does applicant own the								
		d. If applicant does not ov	rn land at point of	diversion, sta	te name and add	ress of owner and	state what step	s have been take	en to obtain right	
		of access:								
	4. F	PURPOSE of USE, AI	MOUNT and S	EASON	d the ensure a	of wester for each a	umore and date	r hatmaan udirl	diversions will	
	2	be made in the table be	ow. Use gallons	per day if rate	is less than 0.0	25 cubic feet per s	second (approxi	mately 16,000 ga	illons per day).	
			r					rock standards (1
				DIRECT	DIVERSION		-	STORAGE		16
		PURPOSE	AMOU	NT -	SEASON OF	DIVERSION	AMOUNT	COLLECT	TON SEASON]
		USE	RATE (Cubic feet per	Acre-feet	Beginning	Ending Date	Acre-feet	Beginning	Ending	
			second or	per year	(Mo. & Day)	(Mo. & Day) .	per year	(Mro. & Day)	Date (Mo. & Day)	
1 26		Irrigation	gailons per day)							
110 18 50	VPC 1	Domestic					- 6			
LAY/+	F1/100	CAMPGRAUNI			1/1	12131				
SALLENI						1 - 1 -				
7	1)									
, hl										
rappmabl		4	1		(TOTAL)			_(TOTAL)		
up to 30 SALYDAY/F		4.0			_ (TOTAL)		-	_(10182)		Į.

					1	year will be	ON AC	RE-FEET	NO	RMAL SEASON	
		CRO	P	ACRES	(Sprinkle	OF IRRIGAT	W11	er year)	Beginning D	ate Endir	g Date
		5-448	7	-	1		1				
	1										
	c.	DOMESTIC: Th	a number of res	idences to be s	erved		tely owned			200	
	b.	OUMESTIC: IT	e total number	of people to be	served 65	Estima	ated daily u	se per pers	on	gallons per day)	
				domestic lawns		1 0000	•		5/day	1100 5	QF1)
						(square fee	t)	(1.0	/ /	
		М	scellaneous do			Dust control a	rea. Numb	er and kind	of domestic an	ımàls, etc.)	
		STOCKWATERIN	G. Kind of S	tock	Ma	ximum Number		. Describ	e type of operat	tion (feed lot, da	ilry, range, et
	d.	RECREATIONA	L: Type of re	creation: Fis	hing 🔲 .	Swimming	□, B	ating [, Other		
		(Submit "Supple	ment to Applica	tioa", form SWI	RCB 1-1, for	justification of	f amount fo	r uses not	listed above.)		*
	DI	ERSION WO						T. Carlo	2	Unmanawas	
	12.	Diversion will b		om		el, reservoir,		charge rati	(cfs/gpd)	Horsepower	
	0			(sump, orrse	1			111	a		
	(b.	Diversion will b	e by gravity by	means of	(1	ipe in unobstr	ucted chann	el, pipe th	rough dam, siph	ion, gate, etc.)	
	6	Estimated total	cost of the dive	rsion works pro	posed is		-			reservoirs, and	main condui
	(c.					ve only cost of	intake, or	neadworks,	, pumps, storage	1 10 301101121	
	(d.)	Main conduit fro	The second secon	1 05	OSS SECTIO	NAL DIMENSI	ON	LENGTH	TOTAL LI	FT OR FALL	(estimated)
	,	CONDUIT Pipe or	MATERIAL (Kind of Pipe	or (i	Pine diamete	or ditch depth bottom width		(feet)	(feet)	(+ or -)	/es(imated)
		channel)	channel linin	g)	and top and	DOLLOW MIGGI		L	(0 -		22 17
		100									
				_				1			
	7	The following	annlies to stora	ee reservoirs: (For reservoi	s having a cap	acity of 25	acre-leet o	r more, complet	te supplemental	form SWRCB
	C.	the iditowing	applies to state		DAM					MESERAOIN	
		Name or num		dam from	Material	Dam	Freeboard height a		Approximate face area when	Approximate	water
		of reservo		level (ft.)	onstruction		spillway cre		full (acres)	(acre-feet)	depth
	-	1									
	()	If water will b	e stored and the	reservoir is no	t at the dive	rsion point, the	maximum i	ate of dive	ersion to offstre	am swrage will	De
_	1		fistream storage	Hill be amount	A brumbiu &	⊴; pan	, .				
	1	Diversion to o								000 000 000 0 <u>000</u>	
	7. F	1 10F 0F 115							in think awaged	in. VEC	ם אם ד
	7. F	LACE OF U	SE.	e the water wil	be used:	YES A	NO	Land is	in joint ownerst	nip: YES	NO [
70000	7. F	LACE OF US	SE s the land when	de their sames	se snaliesati	9ffl ff9\iz bac :	addii catior.	II appilica	mit and a line au	nip: YES	NO □
	7. F	LACE OF US	SE s the land when	e the water will de their names of owner and s	se snaliesati	9ffl ff9\iz bac :	addii catior.	II appilica	mit and a line au	nip: YES] NO □ waterwillb
	7. F	LACE OF US	SE s the land when	de their sames	se snaliesati	9ffl ff9\iz bac :	addii catior.	II appilica	mit and a line au	nip: YES	NO C
	7. F	LACE OF US	SE s the land when	de their sames	se snaliesati	9ffl ff9\iz bac :	addii catior.	II appilica	mit and a line au	nip: YES n land where the	NO C
	. F	LACE OF US	SE s the land when	de their sames	se snaliesati	9ffl ff9\iz bac :	addii catior.	with the o	wher.	a raid miles	
	7. F	Applicant own All joint owne used, give na	SE s the land when rs should inclu ne and address USE IS WITHII	de their names of owner and s	se snaliesati	9ffl ff9\iz bac :	addii catior.	with the o	wher.	IF IRRIGATI	ON ntly cultivate
	7. F	Applicant own All joint owne used, give na	SE s the land wher ors should inclu- ne and address	de their names of owner and s	as applicant tate what arr	and sygn the	application.	with the o	wher.	IF IRRIGATI	ON
	7. F	Applicant own All joint owne used, give na	SE s the land when rs should inclu ne and address USE IS WITHII	de their names of owner and s	as applicant tate what arr	and sygn the	application.	with the o	wher.	IF IRRIGATI	ON ntly cultivate
	7. F	Applicant own All joint owne used, give na	SE s the land where rs should inclu re and address USE IS WITHII 0-acre Subdivis	de their names of owner and s	as applicant tate what arr	and sygn the	application.	with the o	wher.	IF IRRIGATI	ON ntly cultivate
	7. F	Applicant own All joint owne used, give na	SE s the land when rs should inclu re and address USE IS WITHII 0-acre Subdivis I6 of I6 of	de their names of owner and s	as applicant tate what arr	and sygn the	application.	with the o	wher.	IF IRRIGATI	ON ntly cultivate
	7. F	Applicant own All joint owne used, give na	USE IS WITHING-acre Subdivis	de their names of owner and s	as applicant tate what arr	and sygn the	application.	with the o	wher.	IF IRRIGATI	ON ntly cultivate
	7. F	Applicant own All joint owne used, give na	SE s the land when rs should inclu re and address USE IS WITHII 0-acre Subdivis I6 of I6 of	de their names of owner and s	as applicant tate what arr	and sygn the	application.	with the o	wher.	IF IRRIGATI	ON only cultivate
^	7. F	Applicant own All joint owne used, give na	SE s the land when rs should inclu re and address USE IS WITHII 0-acre Subdivis I6 of I6 of	de their names of owner and s	as applicant tate what arr	and sygn the	application.	with the o	wher.	IF IRRIGATI	ON ntly cull

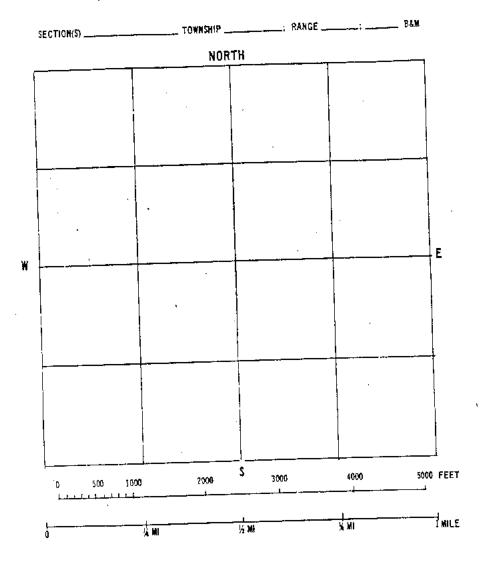
(ATTACH SUPPLEMENTAL SHEETS HERE)

9.	GE	NERAL						معالم الما	(anler.	low	
	ä,	NERAL What is the name	of the post offi	ce most used by	y those livi	ng near the	proposed par	nt of diver	Sibil:	.) VEC F		H Yes
	þ.	Dana and of	44 a alaga #1 1151	e romatiste a Sil	ibaiyision o	Ultic mint	nic arem pack		,,		_ —	***
		state name of Su	bdivision			[[No, 15 SI	ipgralaton or	Meze mun	2 militaribis	104: 120	☐ KC	7
		to it alsocation	individually met	er each service	connection	? YES	<u> </u>	2 (: 11)	Yes, when?		 ,	
£ 🚣	٤.	Once you was sol	Dad the Californ	is Department (al Fish and	Game conc	erning this ph	oposed pro	jecl? TE	., [∟	10 12≤(1	(Y & S , S CB (8
۷.		the Department's	s opinion concer	rning the potent	tial ellects	of your prop	osed project	an fish an	d other will	llife and stat	le measures re	paring
		for mitigation _							··			
	٠.										مامياه	
alignatura Tarak	\$ 7	II No. state the	elfects on fish	and other wildli	ile you fore	see as pote	ntially arising	from you	proposed :	mject	None	
=	9											
	ď	Please name of	ther public agent	cies, il any, fro	on which you	u have obta	ined or are re	dajusą po c	btain appro	vals regardi	ng this projec	t
i.	-	- (H)	of the second		<u> 211 i</u>		<u> </u>					·····
		What are the na		cec of diverters	of water fr	on the soul	ce of supply	do wristie a	n Irom the p	roposed pol:	nt at diversion	1?
	بأرار	N ⊜	Ands and added	1	وأجادته	<i>:</i>		<u>. ·</u>		<u> </u>		· · · · · · · · · · · · · · · · · · ·
ŗ	VU	, <u>, , , , , , , , , , , , , , , , , , ,</u>	1		r.,			<u></u>				
			sed for navigation			ura basals S	ora sienifica:	nt part of c	ach year a	l the paint o	l diversion, or	does
	ſ,	Is the source u	sed for navigation stantially contri	on, Including us	SC Dy Dicase	in used to:	nauleatina in	cluding us	e by pleasa	re boats? _	<u> 8.0</u>	
		the source sub	stantially contri	bute to a water	way walca i	12 H260 IDI	Harricant	J. Barrie				
10.	E.	XISTING WA	TER RIGHT					Castias 1	ves 1	NO C	- -3	
	Do	you claim an ex	isting right for t	he use of all o	part of the	waler sout	ար: By buis api	incation:	(L.) <u>(</u>	± 1,11° €		
		yes, complete tal			year.	. G F	V 6	F 11.	31 0		,	
		,		7	- 5	P. mase	n1			Source	Locati	an cí
		(tibStian' s	of Pights ppropriative.	Year/of First/Use	សូន inc	e made in r luding amo:	ecent years		Season of Use	300120	Point of E	
444)		dwaler.)	7		- 7						
		17 E. 21										
	•			<u> </u>			<u> </u>					÷
11.	A	UTHORIZED) AGENT (U	(ptional)			aa 🗂 Hoosi	matiers d	leci enaled	as foilows: .		
	*	ith respect to:	All matter	s concerning th	is water rig	ntapplicat	.on, ulos:	Matters .	20,31,21010-			
	-					Ad	dress					
	H	ame		 -							een 8 a.m. an	6.5 n.m.l
	-			32 my 2001		- •		(Telep	itione No. 0	i agent perm	GEII O STIIT SIN	3 5 pinner
		s authorized to a										
12.	. 3	(we) declare und	DI REFLICA	*N i		and corre	the best	af my (but)	knowledge	and belief.		
	- 1	(we) declare und	ter penalty of pe	enjury that the a	10045 12 MAI	: Win Chile		• • • • • • • • • • • • • • • • • • • •	_			
											Calilomia	
		Dated		₁₉	, 31							
						Ms. Mr.						
						Miss, Mrs.	•					
						1111 AA 9 1111 34	(Signature of	applicant	(Refer to	Section 67:	of the Board'	s reguiations;
		Il applicants are	members of the	same family								
		tion bushand W	rile, mother, fall	iei, son,		Ms. Mr.						
		brother, sister, e address, please	of reside at	the same		Miss, Mrs.			0.70-1	Section 671	of the Board	's regulations)
		eduicas! hisaac	the land of the land	•			(Şignature o	1 Spoilcan	ri (weier n	. 200,000	-	

Additional information needed for preparation of this application may be found in the leaflet entitled "HOW TO FILE AN APPLICATION TO APPROPRIATE WATER IN CALIFORNIA". If there is insufficient space for answers in this form, attach extra sheets. Please cross reference all remarks to the numbered item to which they may refer. Send application in duplicate to the STATE WATER RESOURCES CONTROL BOARD, DIVISION OF WATER RIGHTS, P. O. Box 2000, Sacramento, CA 95810, with \$100 minimum filing fee.

13. Application Map

(Please complete legibly, with as much detail as possible) (See example in instruction booklet)



- (1) Show location of the spring or stream, and give name,
- (2) Show location of the main ditch or pipe line,
- (3) Indicate clearly the proposed place of use of the water.
- (4) Locate and describe the point of diversion (Less the point at which water is to be taken from the stream or spring) in the following way: Begin at the most convenient known comes of the public land survey, such as a section or quester section corner (if on unsurveyed land more young time survey, such as a section corner, begin at a mark or some natural object or permanent monument that can be readily found and recognized) and measure directly north or south until opposite the point which it is desired to locate; then measure directly east or west in the desired point. Show these distances in figures on the map as shown in the instructions.

14. Environmental Information

An Environmental Information form provided by the State Water Resources Control Board should be completed and attached to this application.

WATER TESTING AND CONSULTING LABORATORY

4237 CARPINTERIA AVE., UNIT 3 CARPINTERIA, CA 93013 (805) 684-3301

DECEMBER 18, 1990

DEPARTMENT OF PARKS & RECREATION GAVIOTA DISTRICT 10 REFUGIO BEACH ROAD GOLETA, CA 93117

WATER ANALYSIS

RE: TUNNEL SPRING WATER
GAVIOTA STATE PARK

ATTENTION: MR. JESS OSBORN

Enclosed are the analytical results of a water sample received from you on December 12, 1990.

This spring water did not pass the standard for general physical analysis due to high concentration of threshold odor number(at 60°C) exceeding the Maximum Contaminant Level(MCL) allowed by the State of California Department of Health in drinking waters. Its general mineral and inorganic chemical analyses including aluminum, passed the MCL. The high odor number was caused by the presence of high concentration of hydrogen sulfide gas(47.8 ppm H2S).

To meet the State Health standard for general physical analysis, it is recommended that the water be treated by chlorination and then filtration using a Granular Activated Carbon filter to remove the sulfur gas which caused the high odor number. An alternative to chlorination is to use a hydrogen peroxide to neutralize the sulfide. The advantage of using the hydrogen peroxide is the fact, it does not form trihalomethane by-products like those of the chlorine.

If there are any questions regarding this report, please feel free to call or write. Thank you.

SINCERFLY.

WATER TESTING AND CONSULTING LABORATORY

4237 CARPINTERIA AVE., UNIT 3 · CARPINTERIA, CA 93013 · (805) 684-3301

CALIFORNIA STATE HEALTH APPROVED LABORATORY

ANALYSIS

REPORT TO: DEPARTMENT OF PARKS & RECE GAVIOTA DISTRICT, 10 REFUG	REATION	LAB NO	901667
GOLETA, CA 93117	DEACH ROA	DATE _	12/18/90
		PAGE_	1of1
ATTENTION: JESS OSBORN		TESTS REQUESTED	GENERAL MINE
SOURCE: TUNNEL SPRING, GAVIOTA STATE		GENERAL PHYSICAL	AND INORGANIC
SAMPLE DESCRIPTION: RAW UNTREATED SPR	ING WATER	CHEMICAL ANALYSES	INCLUDING
DATE COLLECTED: 12/12/90		ALUMINUM AND TOTA	L SULFIDE GAS.
SAMPLE COLLECTOR: JESS OSBORN	1/0		
SAMPLE BOTTLE USED: WTCL i-L plastic; ml ZnAc-NaOH(for H2S): LAB SPLIT, 1/4 L +	1/2 L plasti		1/12/00
wo mi r 0.5 mi nwol(lor nitrate & fluoride)	o.s iii mwa(tor		2/12/90
GENERAL MINERAL ANALYSES	RESULTS	INORGANIC CHEMICAL ANALYSES	
Bicarbonate Alkalinity, mg/1 CaCO ₃	60	Arsenic, mg/1 As (0.05)	RESULTS
Carbonate Alkalinity, mg/1 CaCO ₃	<1	Barium, mg/1 Ba (1.0)	<0.005
Hydroxide Alkalinity, mg/1 CaCO ₃	<1	Cadmium, mg/1 (0.01)	<0.050
Calcium, mg/1 Ca	77	Chromium, mg/1 Cr (0.05)	<0.001
Chloride, mg/1 C1 (500)	180	Lead, mg/1 Pb (0.05)	<0.005
Copper, mg/1 Cu (1.0)	<0.02	Mercury, mg/1 Hg (0.002)	<0.005
Surfactant, mg/1 MBAS (0.5)	<0.02	Nitrate as Nitrogen, mg/1 NO ₃ N (10)	<0.000
ron, mg/1 Fe (0.3)	<0.05	Nitrate as NO ₃ , mg/1 NO ₃ (45.0)	<0.1
Magnesium, mg/1 Mg	12	Fluoride, mg/1 F (2.0)	<0.4
Manganese, mg/1 (0.05)	<0.02	Selenium, mg/1 Se (0.01)	2.0
H factor, unit	7.2	Silver, mg/1 Ag (0.05)	<0.002
Sodium, mg/1 Na	133	ORGANIC CHEMICAL ANALYSES	<0.002
ulfate, mg/1 SO ₄ (500)	229	Chlorinated Hydrocarbons:	
onductance, electrical, 25° C (1600)MMH/CM	1150	Endrin mg/1 (0.0002)	
otal Dissolved Solids, mg/1 TDS(at 180° C)(1000)	644		
otal Hardness, mg/1 CaCO ₃		Lindane, mg/1 (0.004)	
Calcium Hardness, mg/1 CaCO ₃	240	Methoxychlor, mg/1 (0.1)	
Magnesium Hardness, mg/1 CaCO ₃	192 48	Toxaphene, mg/1 (0.005)	
inc, mg/1 Zn (5)	<0.02	Chiorophenois:	
ENERAL PHYSICAL ANALYSES		2,4-D, mg/1 (0.1)	
olor, Unit (15)	15	2,4,5-TP Silvex, mg/1 (0.01)	
urbidity, NTU (5)	15	MISCELLANEOUS	
hreshold Odor Number at 60 C (3)	1.5	Boron, mg/1 B	·
THERS: Theoretical TDS(Summation of Solids)	200	Potassium, mg/1 K	13.4
tal Anions, Milligram Equivalent/liter	703	Sodium Adsorption Ratio (SAR)	3.7
tal Cations, Milligram Equivalent/liter	10.9	Per Cent Sodium	52.7
	11.0	Aluminum(1.0), ppm Al	<0.1
arcland Deviation, Anion-Cation Balance LEGEND: Mg// means milligrams per liter and is also equivalent methods used in testing. Numbers in parenthesis are the Maxim digiting waters.	± 0.4 to parts per million (pr num Contaminant Levi	TOTAL SULFIDES, ppm as H2S	47.8

ing the EPA or State of California standard laboratory procedures. The liability of this laboratory shall not exceed the amount paid for in this report.

REMARKS: This spring water did not pass the standard for general physical analyses due to high concentration of threshold odor number(at 60°C) exceeding the Maximum Contaminant Level(MCL) allowed by the State Health in drinking waters. Its general physical and inorganic chemical

analyses including aluminum, passed the MCL.

t there are any questions, please call or write me personally. Thank you. Sincerely,

WATER TESTING AND CONSULTING LABORATORY

4237 CARPINTERIA AVE. UNIT # 3 CARPINTERIA, CA 93013 Telephone (805) 684-3301

REFERENCE METHODS OF ANALYSIS USED BY THE WATER TESTING & CONSULTING LABORATORY

(BASED FROM EPA 1983 METHODS FOR CHEMICAL ANALYSIS OF WATER & WASTES, AND 1985

STANDARD METHODS FOR THE EXAMINATION OF WATER & WASTEWATER 1644 FOUTTON)

A MENICA ROURTIES Color, Unit Color, Unit Corductance, Microthes/On 25°C Hardness, Total (mg/1 as CaO) Hardness, Total (mg/1 as CaO) The special of the s	STANDARD METHODS FOR THE EXAMINATION OF WATER & WASTEWATER, 16th EDITION).							
Color, Unit Cordictance, Microthos/On 25°C Hardness, Intel(mg/l as GaO) Cotr, Treshold Otor Cit, Unit Cordictance, Microthos/On 25°C Hardness, Intel(mg/l as GaO) Cotr, Treshold Otor Cit, Unit Cordictance, Microthos/On 25°C Cit, Treshold Otor	PARAMETER & UNIT	MEIHOS	REFERENCE					
Credictance, Microtrics/On 25°C Harriness, Total(mg/1 as GaO3 Color, Treshold Other Hi, Unit HESIDIE: mg/1 Total Dissolved (Filterable) Total Suspended(Nm-fiterable) Total Solids Uolatile Jettleable Matter Jett								
### Harriers, Total (ng/l as GaO3 Chr., Theshold Other Comparison with other-free water at 60°C 1		Colocimetric, Platirum-Cobalt	EPA Method 110.2					
Hardness, fotal (ng/l as CaO3 of the control of the		Specific Conductance, Wheatstone Bridge	EPA Method 120.1					
Control Control Control Comparison with other-free water at 60°C 1	₩ Hardness, Total(mg/l as CaCO3		EPA Method 130.1					
Print Fig. 1 Fig. 2 Fi	Odor, The shold Odor							
RESIDE: mg/l								
Total Superdad(Non-fiterable) Gravimetric dried at 103°C-105°C 160.2 160.3 160.3 160.4 160.3 160.4 160.5 160.5								
Total Superiad(Nn-fiterable) Gravimetric dried at 103°-105°C	Total Dissolved (Filterable)	Gravimetric diried at 180°C	" " 160.1					
Total Solides Gravimetric dried at 103°-105°C 160.3 Volatile Gravimetric, Ignited at 550°C 160.4 Settlesble Matter Volumetric, IMFF Grave 160.5 Imperature, % Themmetric 170.1 Imperature, % Themmetric 170.1 Inthidity, NIU Nephemetric 180.1 Malimitum, Al	[] Total Suspended(Non-fiterable)	Gravimetric dried at 103°C-105°C						
Settleable Natter Wolmetric, IMTFF Gare 160.5 Imperature, °C Themmetric 170.1 Imperature, °C Themmetric	[] Volatile							
Tamperature, °C Themmetric " 170.1 180.1 METALS: All Units in mg/1 A Rurrace " 202.2 Arsenic, as As A Rurrace " 205.2 Barium, as Ba A Rurrace " 206.2 Barium, as Ba A Rurrace " 208.2 Beryllium, as Be AA Rurrace " 210.2 Brotn, as B Golorimetric, Outcomin " 210.2 Brotn, as Ca A Rurrace " 213.2 Calcium, as Ca A Rurrace " 215.2 Calcium, as Ca A Rurrace " 215.1 Condit, as Co A Rurrace " 218.2 Hexwalent Orrordim(Cr+6) Chelation Extraction " 218.4 Cromium(Dissolved) A Rurrace " 219.2 Gold, as Au A Rurrace " 219.2 Gold, as Au A Rurrace " 231.2 Iridium, as Ir A Rurrace " 231.2 Iridium, as Ir A Rurrace " 239.2 Magnesium, as Mg A Direct Aspiration " 242.1 Marcury, as Hg Gold Vapor, Sediments " 245.5 Molybernum, as M A Rurrace " 249.1 Marcury, as Hg Gold Vapor, Sediments " 245.2 Palladium, as Rd A Rurrace " 255.2 Ralladium, as Rd A Rurrace " 277.2 Scalium, as Na A Rurrace " 277.2 Scalium, as Na A Rurrace " 277.2 Tin, as S7 A Rurrace " 277.2 Ti	[] Settleable Matter							
B. METALS: All Units in mg/1								
METALS: All Units in mg/1								
Aluminum, Al A. Rumace " 202.2 Arsenic, as As A. Rumace " 202.2 Arsenic, as As A. Rumace " 206.2 Barlum, as Ba A. Rumace " 206.2 Barlum, as Ba A. Rumace " 200.2 Boron, as B A. Rumace " 210.2 Boron, as B Colorimetric, Ourcumin " 210.2 Calcium, as Cd A. Rumace " 213.2 Calcium, as Ca Titrimetric, EDTA " 215.2 Calcium, as Ca Titrimetric, EDTA " 215.1 Curomium, as Cr A. Rumace " 218.2 Calcium, as Cr A. Rumace " 218.2 Calcium, as Cr A. Rumace " 218.3 Cobalt, as Co A. Rumace " 218.3 Cobalt, as Co A. Rumace " 218.3 Cobalt, as Co A. Rumace " 219.2 Calcium, as Ir A. Rumace " 219.2 Calcium, as Ir A. Rumace " 226.2 Iridium, as Ir A. Rumace " 226.1 Lead, as Pb A. Rumace " 226.1 Lead, as Pb A. Rumace " 229.2 Magnesse, as Mn A. Direct Aspiration " 242.1 Manganese, as Mn A. Direct Aspiration " 243.1 Calcium, as Mg A. Rumace " 226.2 Calcium, as Calcium, as Ma. Rumace " 226.2 Calcium, as Calcium, as Ma. Rumace " 226.2 Calcium, as Calcium, as Ma. Rumace " 226.2 Calcium, as Mg A. Rumace " 226.2 Calcium, as Mg A. Rumace " 227.2 Calcium, as	B. METALS: All Units in mg/1		200-1					
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Berium, as Ba								
Beryllium, as Be								
Beron, as B Colorimetric, Ourcumin 11 212.3 Cadmium, as Cd AA Rurrace 123.2 Calcium, as Ca Titrimetric, EDIA 11 215.2 Calcium, as Ca Titrimetric, EDIA 11 215.1 Carcinium, as Ca AA, Direct Aspiration 11 215.1 Carcinium, as Cr AA Rurace 128.2 Hexavalent Ourcumium(Cr+6) Chelation Extraction 11 218.4 Curumium(Dissolved) AA Rurrace 128.3 Cobalt, as Co AA Rurrace 129.2 Cold, as Au AA Rurrace 123.2 Iridium, as Ir AA Rurrace 123.2 Iridium, as Ir AA Rurrace 123.2 Iridium, as Fe AA Direct Aspiration 123.2 Cold, as Au AR Rurrace 123.2 Cold, as Au Cold, as Fe AA Direct Aspiration 123.2 Cold, as Au Cold, as Fe AA Direct Aspiration 123.2 Cold, as Fe Cold Vapor, Manual 123.2 Cold, as Fe Cold Vapor, Manual 124.5 Cold Vapor, Sediments Cold Vapor, Sediments 124.5 Cold, as Ni Cold, as Au Rurrace 124.5 Cold, as Ni Commium as Os AA Rurrace 124.5 Commium as Os AA Rurrace 125.2 Commium, as Rh AA Rurrace 125.2 Cold, as Ni Cold, as N								
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Calcium, as Ca Titrimetric, EDTA " 215.2 Calcium, as Ca AA, Direct Aspiration " 215.1 Calcium, as Ca AA, Direct Aspiration " 215.1 Coronium, as Cr AA Furance " 218.2 Hexavalent Chromium(Cr+ 6) Chelation Extraction " 218.4 Chromium(Dissolved) AA Furance " 218.3 Cobelt, as Co AA Furance " 219.2 Cold, as Au AA Furance " 231.2 Iridium, as Ir AA Furance " 235.2 Iridium, as Ir AA Furance " 236.1 Mercad, as Fb AA Direct Aspiration " 236.1 Mercances, as Mn AA Direct Aspiration " 242.1 Mercance, as Mn AA Direct Aspiration " 243.1 Mercury, as Hg Cold Vapor, Manual " 245.1 Mercury, as Hg (in sediments) Cold Vapor, Sediments " 245.5 Molytchrum, as Mb AA Furance " 246.2 Nickel, as Ni AA Furance " 249.1 Comium as Cs AA Furance " 253.2 Flatium, as Rd AA Furance " 253.2 Fractium, as Rd AA Furance " 253.2 Fractium, as Rd AA Furance " 255.2 Ractium, as Rd AA Furance " 253.2 Scdium, as Rd AA Furance " 273.1 Thellium, as Rd AA Furance " 273.2 Scdium, as Rd AA Furance " 273.1 Scdium, as Rd AA Furance " 273.1 Scdium, as Rd AA Furance " 273.1 Scdium, as								
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Growning Dissolved) A Finnace Cobalt, as Co A Finnace A Finnace Cobalt, as Au A Finnace A Finnace Cold, as Au A Finnace Cold, as Au A Finnace Cold, as Au A Finnace Cold, as Fe A Direct Aspiration Cold, as Fe A Direct Aspiration Cold, as Fe A Direct Aspiration Cold, as Managanese, as Mn A Direct Aspiration Cold Vapor, Manual Cold Vapor, Manual Cold Vapor, Manual Cold, as Ni A Finnace Cold, as Ni Cold, as Ni A Finnace Cold, as Ni A Finnace Cold, as Ni Cold, as Ni A Finnace Cold, as Ni Cold, as Ni A Finnace Cold, as Ni Cold, as Ni Cold, as Ni A Finnace Cold, as Ni Cold,			210.2					
Cobalt, as Co AA Furnace Gold, as Au AA Furnace Balladium, as Rd AA Furnace AA Furnace Balladium, as Rd AA Furnace AA Furnace Balladium, as Rd Balladium, as Rd AA Furnace Balladium, as Rd Balladiu			210.4					
Gold, as At A Furnace " 231.2 I ridium, as Ir A Furnace " 235.2 I Iron, as Fe AA Direct Aspiration " 235.1 I Lead, as Pb AA Furnace " 239.2 Magnesium, as Mg AA Direct Aspiration " 242.1 Manganese, as Mn AA Direct Aspiration " 243.1 Mercury, as Hg Cold Vapor, Manual " 245.1 Mercury, as Hg (in sediments) Cold Vapor, Manual " 245.5 Molytolenum, as Mo AA Furnace " 246.2 Nickel, as Ni AA Furnace " 249.1 Osmium as Os AA Furnace " 252.2 Palladium, as Rd AA Furnace " 255.2 Paltatium, as Rd AA Furnace " 255.2 Rodium, as Rd AA Furnace " 255.2 Rodium, as Rd AA Furnace " 255.2 Selenium, as Se AA Furnace " 255.2 Selenium, as Se AA Furnace " 270.2 Schium, as Na AA Direct Aspiration " 273.1 Thellium, as Tl AA Furnace " 279.2 Tin, as Sn AA Furnace " 279.2 Tin, as Sn AA Furnace " 279.2 Tin, as Sn AA Furnace " 279.2			218.3					
Fridium, as Ir AA Furnace Iron, as Fe AA Direct Aspiration Manganese, as Mn AA Direct Aspiration Mercury, as Hg Cold Vapor, Manual Mercury, as Hg Mercury, as Hg Mercury, as Mo AA Furnace Mercury, as Mo AA Direct Aspiration Mercury, as Hg Cold Vapor, Manual Mercury, as Hg Mercury, as Mo AA Furnace Mercury, as Mo Mercury, as Mo AA Furnace Mercury, as Mo M			219.2					
Iron, as Fe A Direct Aspiration " 236.1 Lead, as Pb AA Furnace " 239.2 Magnesium, as Mg AA Direct Aspiration " 242.1 Manganese, as Mn A Direct Aspiration " 243.1 Mercury, as Hg Cold Vapor, Manual " 245.1 Mercury, as Hg (in sediments) Cold Vapor, Sediments " 245.5 Molybdenum, as Mo AA Furnace " 249.1 Comium as Os AA Furnace " 252.2 Palladium, as Ri AA Furnace " 253.2 Platinum, as Rt AA Furnace " 255.2 Rhodium, as Rh AA Furnace " 265.2 Selenium, as Se AA Furnace " 270.2 Schimm, as Na AA Direct Aspiration " 273.1 Thallium, as TI AA Furnace " 279.2 Tin, as Sn AA Furnace " 279.2			231.2					
Lead, as Pb Lead, as Pb A Furnace Magnesium, as Mg A Direct Aspiration Marganese, as Mn A Direct Aspiration Marganese, as Mn A Direct Aspiration Marcury, as Hg Cold Vapor, Manual Marcury, as Hg (in sediments) Cold Vapor, Sediments Marcury, as Hg (in sediments) Cold Vapor, Sediments Marcury, as Hg (in sediments) Cold Vapor, Sediments Marcury, as Hg Marcury, as Mg Marcury, Marcury Marcury, Marcury			235,2					
Magnesium, as Mg Magnesium, as Mg Magnesium, as Mg Manganese, as Mn Mangan			230,1					
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[] Osmium as Os AA Furnace " " 252.2 [] Palladium, as Pd AA Furnace " " 253.2 [] Platinum, as Pt AA Furnace " 255.2 [] Rhodium, as Rh AA Furnace " 265.2 [] Selenium, as Se AA Furnace " " 270.2 [] Silver, as Ag AA Furnace " " 272.2 [] Sodium, as Na AA Direct Aspiration " 273.1 [] Thallium, as Tl AA Furnace " 279.2 [] Tin, as Sn AA Furnace " 282.2	i i molyodenum, as mo		∠40•∠					
[] Palladium, as Pt AA Furnace " 253.2 [] Platinum, as Pt AA Furnace " 255.2 [] Rhodium, as Rh AA Furnace " 265.2 [] Selenium, as Se AA Furnace " 270.2 [] Silver, as Ag AA Furnace " 272.2 [] Sodium, as Na AA Direct Aspiration " 273.1 [] Thallium, as Tl AA Furnace " 279.2 [] Tin, as Sn AA Furnace " 282.2			249.1					
[] Platinum, as Pt AA Furrace III Rhodium, as Rh AA Furrace III 255.2 AA Furrace III 265.2 AA Furrace IIII 270.2 A Furrace IIII 272.2 A Furrace IIII 273.1 A Furrace IIII 279.2 Thallium, as Ti A Furrace IIII 279.2 A Furrace IIII 282.2	7 7		252.2					
[] Rhodium, as Rh AA Furnace Selenium, as Se AA Furnace AA Furnace Silver, as Ag AA Furnace		AA Furnace	23.2					
Selenium, as Se AA Furrace Silver, as Ag AA Furrace AA Furrace Sodium, as Na AA Furrace		AA Furmace	200.2					
Silver, as Ag AA Furrace " " 272.2 Sodium, as Na AA Direct Aspiration " 273.1 Thallium, as Tl AA Furrace " 279.2 Tin, as Sn AA Furrace " 282.2		AA Flimace	" " 265.2					
Silver, as Ag AA Furrace Sodium, as Na AA Direct Aspiration Thallium, as Ti AA Furrace Tin, as Sn AA Furrace "" 272.2 273.1 279.2 AA Furrace "" 282.2		AA Furnace	" " 270.2					
Sodium, as Na AA Direct Aspiration " " 273.1 [Thallium, as Ti AA Furnace " " 279.2 [] Tin, as Sn AA Furnace " " 282.2		AA Furtace						
[] Thallium, as Ti AA Furnace " " 279.2 [] Tin, as Sn AA Furnace " " 282.2	🗶 Sodium, as Na	AA Direct Aspiration						
[] Tin, as Sn AA Furnace " " 282.2								
	[] Tin, as Sn		" " 282.2					
A birect Aspiration 220.1	M Copper as Oi	AA Direct Aspiration	" " 220.1					

PARAMETERS & UNIT	MERHODS		(Albacatas est	············
METALS Potassium	AA Direct Aspiration		EFERENCE	1.000.4
Titarium, as Ti	AA Furance		EPA Method	
🖈 Zinc, as Zn	AA Direct Acrimation	E,	PA Method	
C. INORGÁNIC, NON-METALLICS (ALL	(hits in me/1)			289.1
[] Acidity as CaCO3	Titrimetric		и н	
Alkalimity as CaCO3			11 II	305.1
Brontide as Br	Titrimetric, pH 4.5			310.1
[] Chloride as Cl	Titrinetric			320.1
Chloride as Cl	Titrimetric, Mercuric Nitrate		11 11	325 . 3
Chlorine as Cl2	Argentanetric Method	1985 St	td. Method	1 407 A
[] Oyanide as ON (Total)	DPD-Spectrophotometric	' E	PA Method	330.5
Fluoride as F	Colorimetric, Spectrophotometric		1 11	335.2
I locitide as I	Colorimetric, SPINS with Distillation	l "	† 11	340.1
	Titrimetric		† †1	345.1
Nitrogen-Amoria	Colorimetric; Titrimetric Distillation	n "	1 11	350.1
Nitrogen-Kjeldshl(Total)	Colorimetric; Titrimetric(Distillatio	n) "	1 11	351.3
Nitrate-N	Colorimetric, Brucine	11	11	352.1
Nitrate-N	Colorimetric-Chromotropic Acid	1985 St	d. Method	418 D
Nitrate N	Cadmium Reduction	1985 St	d. Method	418 C
Nitrate-Nitrite	Colorimetric, Cadmium REduction		A Method	
Nitrite	Spectrophotometric	H	H	354.1
[] Oxygen, Dissolved	Modified Winkler (Full Bottle Technique	" (c	11	360.2
Desphorous(All Forms)	Colorimetric, Ascorbic Acid(Single Re	epent) (11 11	365.3
Silica as SiO ₂ (Dissolved)	Colorimetric	14	*1	370 . 1
Sulfate as SOA	Ilurbidimetric	tt	11	
Sulfide as S= (high HyS)	Titrimetric, Iodine	11		375.4
[] Sulfide as S=(Low HoS)	Colorimetric, Methylene Blue	**	**	376.1
[] Sulfite as SO3	Ti trimetric			376.2
D. ORGANICS (All units in mg/1)				377.1
BOD 5 day, 20°C)	Winkler Azide	71	11	100
[] Chemical Oxygen Demand(CCD)	Titrimetric, Mid-Level	17		405.1
[] Chemical Oxygen Demand(CCD)	Titrimetric, Iow Level	11	•••	410.1
[] Chemical Oxygen Demand(COD)				410.2
Oil and Grease (lotal Recoverable)	Titrimetric, High Level for Saline Wat	ecs		410.3
[] Organic Carbon(Total) - TCC	Gravimetric, Separatory Furnel Extract Confustion-Infrared	ion "		413.1
Petrolam Hydrocarbons, Total				414.1
[] Fhemolics, Total Recoverable	Freen Extraction- Silica Gel	1985 S	td. Methox	±s 503 E
Methylene Blue Active Substances (MBAS)	Spectrophometric, Mercal 4-AAP-Distill	ation E	PA Metinod	420.1
[] Pesticides	CONTRIBUTE, LEGISTERS RITE		" "	425.1
Herbicides	Gas Chromatographic	1985 S	td. Method	1 p.538
	•	" (17 19	p.550
E. RADIOLOGICAL:				•
Gross Alpha, pCi/1	Proportional Counter	** ,	1 (1	615
Gross Beta, pCi/l	11 11		1 11	p. 640
F. TOXICITY BIOASSAY			••	p. 640
	96-Hr. Static Bioassay	11)	1 11	***
-225	Total Didasay		••	p. 690
OTHERS:				
OTIUA/O.				

WATER TESTING AND CONSULTING LABORATORY

4237 CARPINTERIA AVE. UNIT # 3
CARPINTERIA, CA 93013
Telephone (805) 684-3301

MAXIMUM CONTAMINANT LEVELS(MCL)

The Maximum Contaminant Levels(MCL) presented here are in accordance with the State of California Health and Safety Code, Title 22. When your water contains constituents exceeding Maximum Contaminant Levels as indicated below, local or state health authorities should be notified.

GENERAL MINERAL ANALYSIS

II. GENERAL PHYSICAL ANALYSIS

Color, Unit	15
Turbidity, NTU	5
Odor Number, (Threshold), 60°C	3

III. INORGANIC CHEMICAL ANALYSIS

Arsenic, mg/1 As	1.0 0.010 0.05 0.05 0.002 10.0 45.0 0.01
Silver, mg/1 Ag	0.05

*(For Santa Barbara County, MCL for Fluoride is 2.0 ppm)

IV. ORGANIC CHEMICAL ANALYSIS

Chlorinated Hydrocarbons (Pesticides) -\$120.
Endrin, mg/1 0.0002
Lindane, mg/1 0.004
Methoxychlor, mg/1 0.1
Toxaphene, mg/1 0.005
Chlorophenoxys (Herbicides)- \$120.
2,4-D, mg/1 0.1
2,4,5,-TP (Silvex), mg/1 0.01

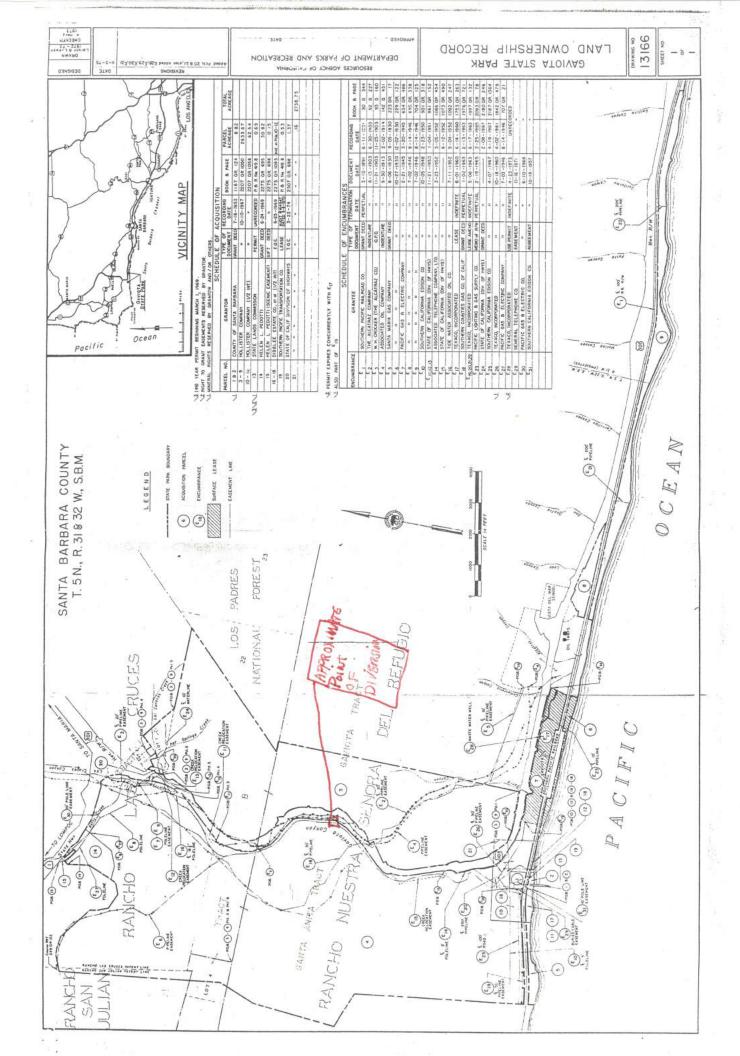
**V. NATURAL RADIOACTIVITY

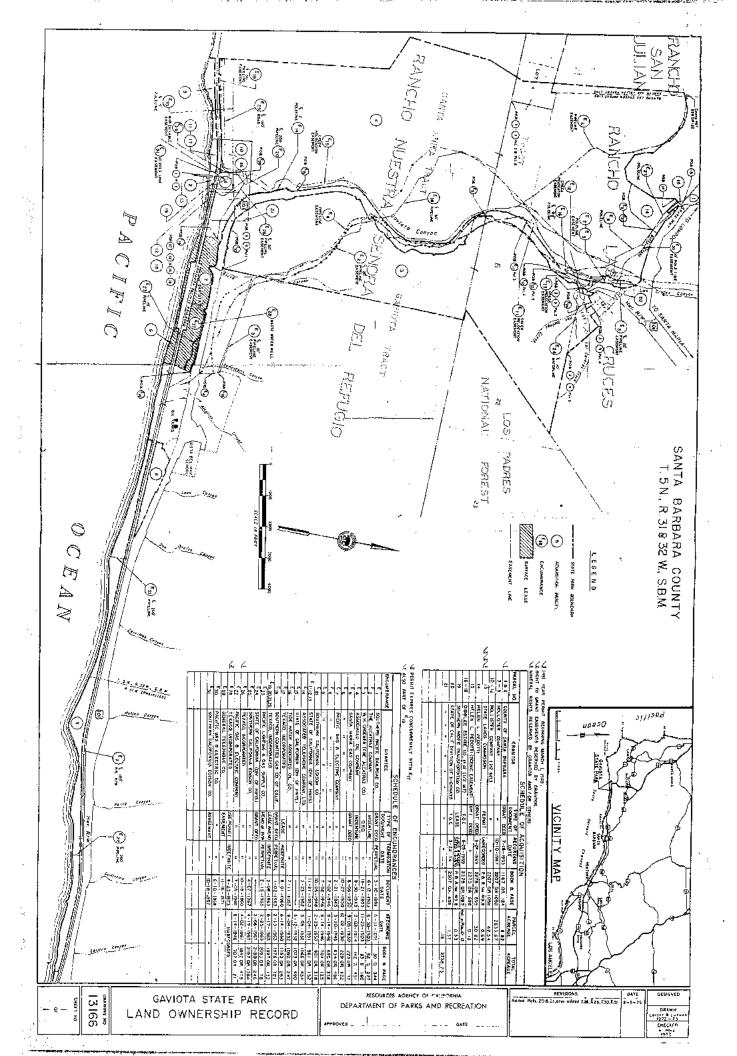
GROSS	BETA,	pCi/1	 50.0

* Note Dependent on the annual average of maximum daily air temperature: 1.4 mg/1 for $\overline{79.3}$ - $79.5^{\circ}F$; 1.6 mg/1 for 70.7 - $79.2^{\circ}F$; 1.8 mg/1 for 63.9 - $70.6^{\circ}F$; 2.0 mg/1 for 58.4 - $63.8^{\circ}F$; 2.2 mg/1 for 53.8 - $58.3^{\circ}F$ and 2.4° mg/1 for $53.7^{\circ}F$ and below.

** Note: Gross Alpha for combine Radium 226 and Radium 228; Gross Beta for Gross Alpha particle activity including Radium 226 but excluding Radon and Uranium.

**CALIFORNIA STATE HEALTH APPROVED LABORATORY*





7.0 WATER RESOURCES TECHNICAL REPORT

Regional Hydrologic Setting

The site is situated within the Central Coast Hydrologic Study Area (CCHSA) as defined by the California Department of Water Resources (CDWR, 1981). The CCHSA encompasses approximately 56 groundwater basins and areas of groundwater storage in the central coast portion of California from Santa Cruz on the north to Carpinteria on the south. In general, the CCHSA has a Mediterranean type climate characterized by a mild to moderate rainy season during winter months followed by a dry season with prevailing mild to very hot temperatures during spring, summer and fall. Drainage in the area is generally southward and westward toward the Pacific Ocean.

The project site is located in the west Santa Barbara coastal hydrologic area, a subdivision of the CCHSA. This area extends from Point Arguello to Ellwood and from the crest of the Santa Ynez Range south and southwest to the sea. The area includes the watershed of Gaviota Creek, one of a few streams in the area that extends north of the crest of the range. The west Santa Barbara coastal hydrologic area is characterized by south and southwest flowing streams which drain the southern flank of the east-west-trending Santa Ynez Range. Groundwater in the area is typically present in Tertiary sandstone bedrock and shallow alluvial aquifers along the valley floors (Miller and Rapp, 1968). No major unconsolidated groundwater reservoirs are present in the area.

The climate in the west Santa Barbara coastal hydrologic area is semi-arid, Mediterranean-type. Summers are generally cool and dry, and winters are mild. Precipitation occurs as rainfall and is variable in the area, averaging about 16 inches per year near the coast to over 30 inches per year on the high mountain slopes of the Santa Ynez Range (Santa Barbara County Flood Control District, 1986). The wettest months are usually December through March. Other forms of precipitation in the area include snow which falls occasionally on the highest slopes and coastal fog which is especially common in the summer months.

Site Hydrologic Conditions

Climate

The climate at the site is similar to that described for the west Santa Barbara coastal hydrologic area (see Regional Hydrologic Setting). Rainfall at the site averages approximately 20 inches per year based on data from rain gauges at Gaviota State Park and Nojoqui Falls County Park (Santa Barbara County Flood Control District, 1986). Although specific records are not available regarding wind at the site, residents and visitors to the area report that strong westerly and northerly winds are common in the area. The Gaviota Gorge area is noted for its down canyon winds which are especially strong in the morning hours. Fog is also common in the area, although the coastal fog typical of the summer months often does not reach the site or burns off earlier than along the coast.

Surface Water

Springs

This section presents a discussion of several springs in the vicinity of the site. These springs include the roadcut springs north of Highway 1, the Pedotti spring, Gaviota hot and cold springs, and Tunnel spring. The locations of these springs are shown on Figure 7-1.

Roadcut Springs

Several springs are present along roadcuts on the north side of Highway I about 1/4 mile west of the San Julian Road turnoff. These springs apparently emanate from the Matilija Formation along a splay off the North Branch Santa Ynez fault (see Geology Technical Report for discussion of faults). Flow from at least three of the larger springs is captured by 2-inch diameter pipes driven into the slope and routed to the base of the slope. Here, the flow discharges into an open channel and eventually flows through a culvert under the highway to the West Fork Gaviota Creek. This collection system was probably installed by Caltrans at the time of road construction to reduce the potential for bank erosion and landslides. Inspection of the discharge pipe in August 1986 indicated a total flow of about I gallon per minute (gpm). The water quality is unknown, but is likely high in mineral content as evidenced by the accumulation of white powdery precipitate near most of the springs. The location of most of the springs is marked by the presence of willow trees.

Pedotti Spring

A spring located 1/4 mile west of the Pedotti ranch house (Figure 7-1) formerly supplied water for domestic and agricultural use, although the spring is no longer used for such purposes (Mr. A. Pedotti, personal communication, 1986). The spring apparently emanates from the Matilija Formation and may represent the location of a trace of the North Branch Santa Ynez fault. The yield of the spring is reported to be several gpm of highly mineralized water.

Gaviota Hot Spring

Gaviota Hot Spring, also called Las Cruces Hot Spring, is located about 1/2 mile east of the site. The spring is located on State Park property and attracts many visitors annually (California Department of Parks and Recreation, 1973). About 25 gpm of 85 degrees (F) water flows from the spring into two pools (Terres, 1984). The spring emanates from landslide or alluvial fan debris which covers the South Branch Santa Ynez fault (Terres, 1984; Dibblee, 1950).

Gaviota Cold Spring

About 600 feet northwest of Gaviota Hot Spring is another spring which supplies water for the State Park residences at Las Cruces. Yield from the spring is reported to be variable (Bill Howard, State Department of Parks and Recreation, Personal Communication, 1986). About 10 gpm is used from the spring with any overflow going to Hot Spring Creek. Water from the spring reportedly contains about 18 to 20 parts per million (ppm) hydrogen sulfide (H2S) and requires acration prior to domestic use.

Tunnel Spring

Tunnel Spring is located within the walls (east side) of the tunnel on the east side of Highway 101 (Department of Parks and Recreation, 1973). A pipe reportedly carries the spring flow westward to Gaviota Creek.

Streams

Several streams are present in the vicinity of the site as shown on Figure 7-1. These are the West Fork Gaviota Creek, Gaviota Creek, Las Cruces Creek, Hot Spring Creek and Las Canovas Creek. The site is located just upstream from the confluence of the West Fork Gaviota Creek and Las Cruces Creek. The only stram gauge in the area is operated by the U.S. Geological Survey and is located on Gaviota Creek about 1 mile south of the site.

The West Fork Gaviota Creek (also called Las Nutrias Creek) generally flows east and south. Stream flow in some parts of the creek is perennial, although no flow was observed in the vicinity of the site during August, 1986. It is likely that surface flow during the summer months is present only where bedrock underlying alluvium is shallow which would tend to cause surface discharge of underflow through the alluvium.

Las Cruces Creek drains the Canada de Las Cruces watershed and flows southerly to its confluence with upper Gaviota Creek just north of the Highway 1 crossing. The confluence of these two creek was originally farther south but was modified to its present location during construction of the highway interchange. Flowing water was observed in the streambed at the Highway 1 crossing during August, 1986.

Las Canovas and Hot Spring creeks are minor drainages on the west flank of Gaviota Peak which converge just east of Highway 101 and empty into Gaviota Creek. Flow near Gaviota Creek is probably perennial due to the presence of Gaviota hot and cold springs.

Flow from all the above mentioned creeks eventually empties into Gaviota Creek which flows southward through Gaviota Gorge to the sea. Gaviota Creek is one of the major streams on the Santa Barbara coast and has an estimated average annual baseflow in the vicinity of the tunnel of about 2.3 cfs (Miller and Rapp, 1968) although stream gauge records indicate that periods of no flow may occur during dry years (U.S. Geological Survey, 1985). The highest recorded flow at the gauge was about 5,270 cfs which occurred in January, 1983.

Ground Water

Hydrogeologic Units

Hydrogeologic units in the site vicinity can be broadly divided into two groups having dissimilar water-bearing characteristics: (1) Tertiary age bedrock units, particularly sandstones; and, (2) Quaternary age unconsolidated units. The bedrock units generally form the hills and mountains in the area and underlie the unconsolidated units at

depth. Unconsolidated units generally consist of alluvium which is present to variable depths in the site vicinity and in the nearby valley floors.

Bedrock units in the area that yield or are suspected of yielding water to wells are the Matilija, Gaviota, Alegria and Vaqueros formations (Hoffman and Associates, 1986a; Hoover & Associates, 1986a). These units are composed of primarily sandstone or siltstone and are described in detail in the Geology Technical Report. The distribution of these units in the area is depicted on Figure 10-3 of the Geology Technical Report.

Unconsolidated water-bearing deposits in the area are limited to alluvium along major creek bottoms. Alluvial deposits consist of various mixtures of clay, silt, sand and gravel of varying thickness. Several wells are present in the area which penetrate alluvial deposits.

Hydraulic Properties

Little information exists regarding the hydraulic properties of hydrogeologic units in the area because of the small number of wells penetrating the units and limited availability of data. Figure 7-1 shows the location of known wells in the site vicinity. The pertinent well characteristics and hydraulic properties of water bearing units are summarized on Table 7-1. A discussion of the characteristics of each formation is provided below.

Matilija Formation

The Matilija Formation is penetrated by the Las Cruces Well #3 and possibly to a very limited extent by the Pedotti Well (Hoover & Associates, 1986b). The Las Cruces Well #3 penetrated 170 feet of blue-gray, fine to coarse grained water bearing sandstone (Hoover & Associates, 1986b). The presence of dense, well cemented sandstone in outcrops of the formation suggests that groundwater is contained in fractures rather than in interstitial pores. A 24-hour aquifer test was performed on the Las Cruces Well #3 at a rate of 30 gpm for 24 hours which produced about 80 feet of drawdown in the well. The transmissivity of the Matilija Formation was calculated to be about 220 gallons per day per foot of aquifer (gpd/ft).

Periodic monitoring of the Pedotti Well (which may or may not penetrate the Matilija Formation) during the 24-hour aquifer test showed no change in water levels within the well (Paul Sorenson, Hoover & Associates, personal communication, 1986). The Pedotti Well is located about 1,000 feet northwest of the Las Cruces Well #3 and is reported to be 132 feet deep (Hoover & Associates, 1986a). The lack of response in the Pedotti Well may indicate that the well is either not completed in the Matilija Formation, the Matilija Formation has an extremely low storage coefficient or that a barrier (fault) separates the wells. It is suspected that the Pedotti Well is bottomed in shale rather than Matilija Formation because of the presence of abundant clay in the well when it was bailed during pump overhaul (Rick Hoffman, Hoffman and Associates, personal communication, 1986). This clay may represent decomposed shale (Anita Formation) bedrock.

A 72-hour aquifer test was also conducted on the Las Cruces Well #3 (Hoover and Associates, 1986c). The well was initially pumped at a rate of 20 gpm which apparently produced excessive drawdown at the end of 2.83 days of testing. The flow

depth. Unconsolidated units generally consist of alluvium which is present to variable depths in the site vicinity and in the nearby valley floors.

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Little information exists regarding the hydraulic properties of hydrogeologic units in the area because of the small number of wells penetrating the units and limited availability of data. Figure 7-1 shows the location of known wells in the site vicinity. The pertinent well characteristics and hydraulic properties of water bearing units are summarized on Table 7-1. A discussion of the characteristics of each formation is provided below.

Matilija Formation

The Matilija Formation is penetrated by the Las Cruces Well #3 and possibly to a very limited extent by the Pedotti Well (Hoover & Associates, 1986b). The Las Cruces Well #3 penetrated 170 feet of blue-gray, fine to coarse grained water bearing sandstone (Hoover & Associates, 1986b). The presence of dense, well cemented sandstone in outcrops of the formation suggests that groundwater is contained in fractures rather than in interstitial pores. A 24-hour aquifer test was performed on the Las Cruces Well #3 at a rate of 30 gpm for 24 hours which produced about 80 feet of drawdown in the well. The transmissivity of the Matilija Formation was calculated to be about 220 gallons per day per foot of aquifer (gpd/ft).

Periodic monitoring of the Pedotti Well (which may or may not penetrate the Matilija Formation) during the 24-hour aquifer test showed no change in water levels within the well (Paul Sorenson, Hoover & Associates, personal communication, 1986). The Pedotti Well is located about 1,000 feet northwest of the Las Cruces Well #3 and is reported to be 132 feet deep (Hoover & Associates, 1986a). The lack of response in the Pedotti Well may indicate that the well is either not completed in the Matilija Formation, the Matilija Formation has an extremely low storage coefficient or that a barrier (fault) separates the wells. It is suspected that the Pedotti Well is bottomed in shale rather than Matilija Formation because of the presence of abundant clay in the well when it was bailed during pump overhaul (Rick Hoffman, Hoffman and Associates, personal communication, 1986). This clay may represent decomposed shale (Anita Formation) bedrock.

A 72-hour aquifer test was also conducted on the Las Cruces Well #3 (Hoover and Associates, 1986c). The well was initially pumped at a rate of 20 gpm which apparently produced excessive drawdown at the end of 2.83 days of testing. The flow

rate was then reduced to 10 gpm for the remainder of the test. Hoover and Associates (1986c) calculated the transmissivity of the Matilija Formation from recovery data gathered during the aquifer test. They estimated the transmissivity of the aquifer to be 2,574 gpd/ft and 11 gpd/ft respectively for the deeper and shallower parts of the aquifer. This extremely wide range in transmissivity values may reflect varied flow rates during the test.

Gaviota Formation

The Gaviota Formation is apparently penetrated by two wells and a test hole in the area. Both the Caltrans Well and the Las Cruces Well #1 (Figure 7-1) are completed in both the alluvium and Gaviota Formation. A 24-hour aquifer test was performed on the Las Cruces Well #1 by Hoffman and Associates (1986b) at a flow rate of 10 gpm resulting in about 38 feet of drawdown. The aquifer transmissivity was calculated at 660 gpd/ft. Hoffman and Associates (1986b) indicated that the transmissivity value was relatively high for bedrock aquifers in the area and suggested that all or nearly all the water entering the well emanated from the alluvium rather than the Gaviota Formation. Review of records regarding the Caltrans Well (Hoffman and Associates, 1986a) indicated that it probably produced little or no water from the Gaviota Formation.

A test hole drilled through alluvium into the Gaviota Formation at a location south of the site penetrated 550 feet of claystone, siltstone and sandstone of the Gaviota Formation (Hoffman and Associates, 1986c). Although the overlying alluvium was water bearing, the Gaviota Formation was judged to be non or very slightly water bearing on the basis of cuttings analysis and electric log interpretation. The Gaviota Formation portion of the test hole was, therefore, abandoned.

On the basis of the information presented above, it is inferred that the Gaviota Formation in the site vicinity has a very low potential to yield water. Results of aquifer tests on wells partially completed in the Gaviota Formation apparently do not reflect the hydraulic properties of the unit.

Alegria Formation

The State Park and Chevron-Alegria Well penetrate the Alegria Formation south of the site (Hoover & Associates, 1986a; Bill Howard, State Department Parks and Recreation, personal communication; Westec, 1985). Production from the State Park Well is about 30 gpm; production from the Chevron-Alegria Well is reported to be 60 gpm. An aquifer test conducted on the Chevron-Alegria well indicates an aquifer transmissivity of 592 gpd/ft Westec, 1985).

Vaqueros Formation

No wells are known to penetrate the Vaqueros Formation in the site vicinity. A test hole drilled on the Hanson property about 1/2 mile north of the site encountered about 430 feet of clay and sandstone assigned to the Vaqueros Formation (Hoover & Associates, 1986d). Water production during drilling was extremely low (1-2 gpm) and the test hole was subsequently abandoned. Consequently, the hydraulic properties of the aquifer in this area are unknown.

Alluvium

Several wells penetrate alluvium in the site vicinity including the Las Cruces Wells #1 and #2, Caltrans Well and Pedotti Well (Figure 7-1, Table 7-1). Most alluvial wells in the area are reported to yield about 10 gpm (Hoover & Associates, 1986a). Aquifer tests (24-hour) were performed on the Las Cruces Wells #1 and #2 which yielded transsmissivity values of 660 and 2200 gpd/st (Hoffman and Associates, 1986c and 1986b). Although these values suggest a wide range of aquifer characteristics, it is possible that the age and poor condition of the Las Cruces Well #1 may have affected the test results. A 72-hour aquifer test was also performed on the Las Cruces Well #2 (Hoover & Associates, 1986c). The flow rate during the test was initially 10 gpm, but was reduced to 5 gpm after 7 hours due to excessive drawdown. Transmissivity of the aquifer, calculated from recovery data from the 72-hour test, is 181 and 907 gpd/st respectively for deep and shallow parts of the aquifer (Hoover and Associates, 1986c). Therefore, it appears that the transmissivity of the alluvial aquifer is high variable.

Ground Water Quality

Selected analyses of ground water from wells in the site vicinity are summarized on Table 7-2. Water quality data are available for alluvium, and the Matilija and Alegria formations. In general, groundwater in the area is of moderate to poor quality with alluvial groundwater characterized by slightly better quality better quality with alluvial groundwater characterized by slightly better quality than groundwater in the bedrock aquifers. The concentration of total dissolved solids (TDS) in groundwater from alluvial or suspected alluvial wells range from 912 to 1703 parts per million (ppm). The high range in values may be due to partial mixing with poorer quality bedrock groundwater which may be entering some of the wells. The Las Cruces Well #2 reportedly penetrates only alluvium and produces water with a TDS of 1057 ppm (Hoffman and Associates, 1986c). Alluvial water collected from the Las Cruces Well #2 during 24 and 72 hour aquifer tests showed approximately the same quality. The water exceeds the State Health drinking water standards for manganese and turbidity. Hydrogen sulfide (H₂S) analysis is available only for the Las Cruces Well #2 and indicates that H₂S is present in low concentration in the alluvium.

Groundwater in the Matilija Formation was analyzed after 24 and 72 hour aquifer tests (Table 7-2). Analyses showed an overall decline in water quality from the 24 to the 72 hour test. The water has a TDS of 1280 ppm with a relatively high H₂S content (31.6 ppm). The water exceeds State Health drinking water standards for sulfate, electrical conductance and TDS.

Groundwater in the Alegria Formation is of poor quality with TDS concentrations ranging from 1674 to 2898 ppm (Table 7-2). Alegria Formation groundwater also has a relatively high H₂S and iron content.

Groundwater Use

Two wells currently extract groundwater in the area. These are the Pedotti and State Park wells (Figure 7-1). Water from these wells is used primarily for domestic purposes. The Pedotti Well extracts water from the alluvial aquifer (See Hydraulic Properties) for use at the ranch house and surrounding buildings. Annual yield of the well is unknown but likely does exceed several acre feet per year (AFY).

The State Park well extracts water from the Alegria Formation for use at Gaviota State Park campground. Annual yield of this well is unknown.

In addition a third well, Chevron-Alegria, which has been drilled and tested but is not presently operating is located in the area, This well will extract water from the Alegria Formation, to be used for the Chevron Gaviota processing facility.

Groundwater Recharge and Discharge

Groundwater within the bedrock aquifers in the site area is recharged primarily by direct infiltration of rainfall and runoff, and possibly by seepage from overlying alluvial aquifers (Miller and Rapp, 1968). Additional recharge may occur through underflow of groundwater from adjacent bedrock aquifers. Groundwater in alluvium is recharged primarily through direct infiltration of rainfall and runoff, by discharge of bedrock aquifers and springs, and possibly to a minor extent by percolation of irrigation/septic water. Because of the general lack of well and aquifer data in the area including history of use information, quantitative estimates of the amount of annual recharge to the aquifer systems can not be made.

The majority of groundwater discharge from bedrock aquifers in the site area is through base flow or low flow in streams and underflow in alluvium, and possibly by underflow into adjacent bedrock aquifers. Bedrock aquifers in the area may also discharge groundwater through springs, evapotranspiration and by well pumpage (Miller and Rapp, 1968). Groundwater in alluvium is discharged by streamflow, evapotranspiration, infiltration into underlying bedrock aquifers and well pumpage. Quantitative estimates of discharge from aquifers in the site are not available.

Existing Flood Hazards

The West Fork Gaviota Creek has experienced several moderate floods in recent years. The United States Geological Survey (USGS) stream gauge located on Gaviota Creek about 1 mile downstream of the site has recorded the peak flows shown in Table 7-3. These floods have not caused any significant damage near the site except for stream bank erosion that has partially underminded San Julian Road in several locations.

Peak flows for the 100-year flood at the project site were determined based on measured stream flows at the USGS stream gauge on Gaviota Creek. Table 7-4 shows the peak flow and recurrence interval for floods at the Gaviota Creek stream gauge calculated using Log-Pearson Type III frequency analysis. The 100-year peak flow calculated for the gauge site was then transferred to the project site using the peak flow relationship:

$$Q_1 = Q_2 (A_1/A_2)^{0.70}$$

Where:

Q1 = peak flow at the project site;

Q2 = peak flow at the gate site;

A1 = drainage basin area at the project site; and

 $A_2^{\frac{1}{2}}$ = drainage basin area at the gauge site.

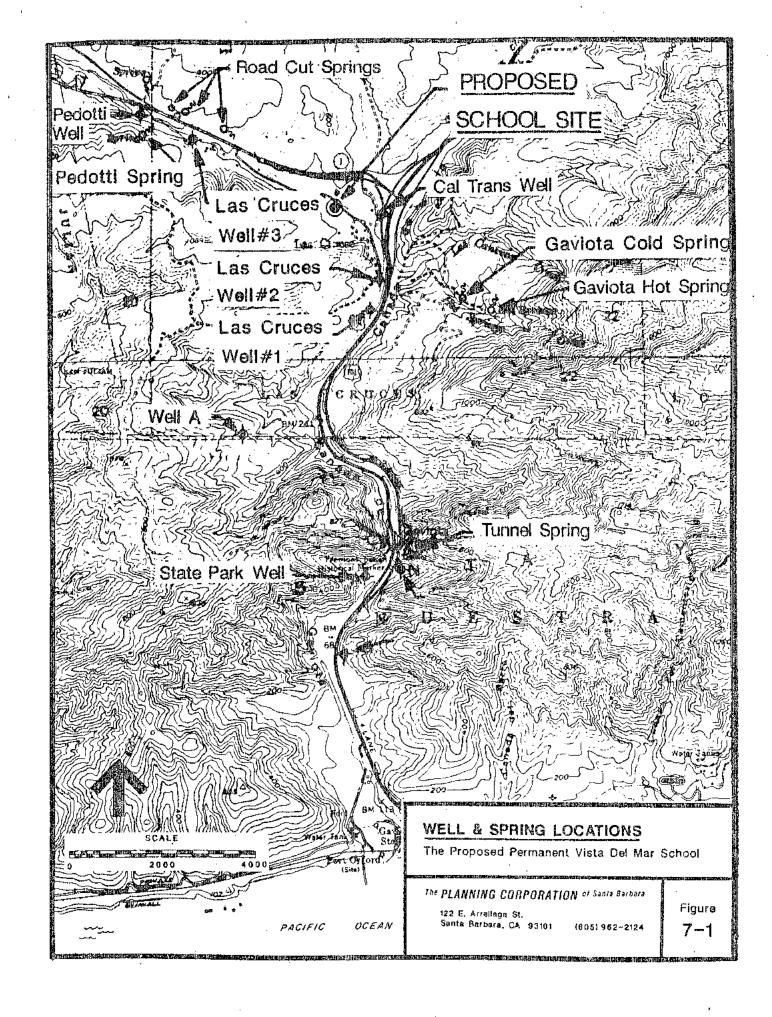
The exponent value (0.70) was obtained from regional investigations completed by Santa Barbara County Flood Control District (personal communication, Will Copeland, SBCFCD). Estimated peak flows at various points near the project site are reported in Table 7-5.

The computer program HEC-2 was used to calculate the 100 year flood profile for Gaviota Creek and the West Fork Gaviota Creek in the vicinity of the project. The program uses a standard step backwater procedure that applies Bernoulli's Theorem at cross sections along the stream and Manning's equation for headloss between cross sections.

Cross sections were compiled from a combination of topographic maps made using aerial photography and field surveys where heavy brush prevented accurate mapping from photographs. Mapping was done at a scale of 1"=40' with 2 foot contour intervals. Field surveying was done to 1/2 foot elevation precision.

Manning's n values were estimated from field inspections and published guides. Channel n values ranged from 0.06 to 0.07. Overbank n values ranged from 0.09 to 0.10. The time of concentration is nearly the same for both the West Fork Gaviota Creek and Gaviota Creek at the project site. Therefore it was assumed that the peak flow in both creeks occurs at the same time. Starting water surface elevations for the computer modeling were based on peak flows.

Figure 7-2 shows the location of the 100 year flood plain along the West Fork Gaviota Creek in the site vicinity calculated using the HEC-2 computer program. Table 7-6 presents the elevation of the water surface at various locations along the creek. All project facilities are outside the 100 year plain. The only existing structures within the 100 year flood plain are the casterly of the two adobe buildings north of the creek, portions of San Julian Road and several bridges in the site vicinity.



The Proposed Permanent Vista Del Mar School Figure 7-2 The PLANWING CORPORATION of Santa Barbara 122 E. Arrellage St. Santa Barbara, CA 93161 (805) 962-2124 100 YEAR FLOOD PLAIN 100 Year Flood Plain Boundary Line

TABLE 7-1

WATER WELL AND AQUIFER CHARACTERISTICS LAS CRUCES AREA

	Well Name	Depth of Completed Well (feet)	Producing Aquifer	Casing Diameter (inches)	Perforated Interval	Yield (GPM)	Aquifer Transmissivity (GPD/ft)
	Las Cruces Well #1	260	Alluvium	.=	*** *** *** *** *** *** *** *** *** **	01	660
	Las Cruces Well #2	60	Alluviam	. 8	30 - 50	10	2200 (24-hour) 181-907 (72-hou
	Las Gruces Well #3	230	Matilija	8	100 - 220	2.2	220 (24-hour) 11-2574 (72-hou
	Pedotti Well	132	Alluvium	-		10	(m
	CalTrans Well	148	Alluvium/ Gaviota?	. 8	40 - 148	9	-
أوني	State Park Well	485	Alegria	8	-	30	· -
	Chevron Alegria Well	650	Alegria	\ 8	M.A.	60	592
	Well A	-	Gaviota?	- -		<3	1 -
	Well B	-	Alegria?	-	brys		-
	•						

\$4 44 0	ŀ	figh Turbid££y	Gerco G.4	Migh Turbldity	G.18 pps Yery high Tunbidity	Boron 1.5ppm Migh solor, cast and turbidity	Boron O. 19a/1 Boron G. 7	î
Hanganese (αc/1)	<0.03	0.18	6,09	\$\$** 0	0.18 pps		G 8	ı
Fluoride (Eg/1)	0.6	0.75	0.68	, , ,	6.6 pp:	1,65	88.5	,
Celcium (eg/1)	314	28	ĸ	₩.	216	ដ	ធ	•
Hegnesium (Ag/1)	55	. 97	8 2	77	*	92	53	1
Sedius (ng/1)	100	226	0.01	210	114	524	490	ı
Potecasum (mg/l)	\$	n	erv, ç çı d	ut.	ı	3.8	5.	ı
Carbonate (mg/1)	Mone	None	₽	Hona detected	Rong delected	g .	5	j
Bicatbonate (mg/1)	451	439	33.	433	397	560	500	ı
Chlorides (nsg/1)	82	8 59	11		ox t	290	170	
Sulfate (mg/l)	969	247	266	198	403	. 707	547	ŧ
Iron (ng/1)	9,4	0.2	0.13	1.7	1.8 pp=	0.02		ı
N25 (ppa)	· ·	alight odor (not analyzed)	8 ::	wiight to My oder (met analyzed)	<u>.</u>	Ji.6 (atzong odor)	Mat analyzed	22 (strong odor)
Totel Dissolved Solids (ppm)	1703	1057	912	256	1302	1280	1692	2858
Hell Nene	Pedoili Ks.11	tas Gruced Mell #2 (24-hour test)	Les Couces Hell \$2 (72-hour tect)	Las Cruces Well fi	Csi T ra na Hall	Las Cruces - Well #3 (20-hour test)	Las Cruces Kell 33 (72-hour test)	State Park Wall
19 J 1777 Y	. Alluvium/Anita(1), (2)	Alluvium[J]	Alluvium (6)	Aliuvium/Gevicta(?)(1),(5)	Alluvium/Nayiçt _s (1),(6)	Matilija Formation(7)	Metilije Formation(4)	Alegrie Formation(8)

物質 ...

Noxious ador, wery hard

Other

Mariganosa (mg/1) 0.33

Well probably derives all or most of yield from alluvium.

Moffmen and Associates, 1986d. Moffmen and Associates, 1986s.

Moover and Associates, 19865.

Hoffman and Associates, 1986b. Floyd V. Wolls, Inc., 1985. Hoover and Associates, 1986c. Sill Howard, Stats Department of Parks and Recreation, Personal Communication. Waster, 1985. ~ 6 6

HISTORIC PEAK FLOOD FLOWS GAVIOTA CREEK NEAR GAVIOTA, CALIFORNIA USGS GAUGE NO. 11120550

Date	Peak Flow (cfs)	Recurrance Interval (years)
1/23/83	5220	17
1/24/67	4000	10
1/18/73	3940	6

Notes: 1. Period of Record 1967-1985

2. Recurrance interval based on a Log-Pearson type III analysis.

CALCULATED PEAK FLOOD FLOWS GAVIOTA CREEK NEAR GAVIOTA, CALIFORNIA USGS GAUGE NO. 11120550

Annual Exceedance Probability	Recurrance Interval (years)	Peak Flow (cfs)
0.50		
	2	1,005
0.20	5	3,034
0.10	10	•
0.02	• •	5,130
	50	11,868
0.01	100	15.571

Notes: 1.

- l. Period of Record 1967-1985
- 2. Generalized skew (-0.30) and systematic record skew (-0.538) used to develop a weighted skew (-0.413). Peak flows are based on the weighted skew.
- 3. Annual exceedance probability based on Log-Pearson Type III analysis.

100-YEAR PEAK FLOOD FLOWS

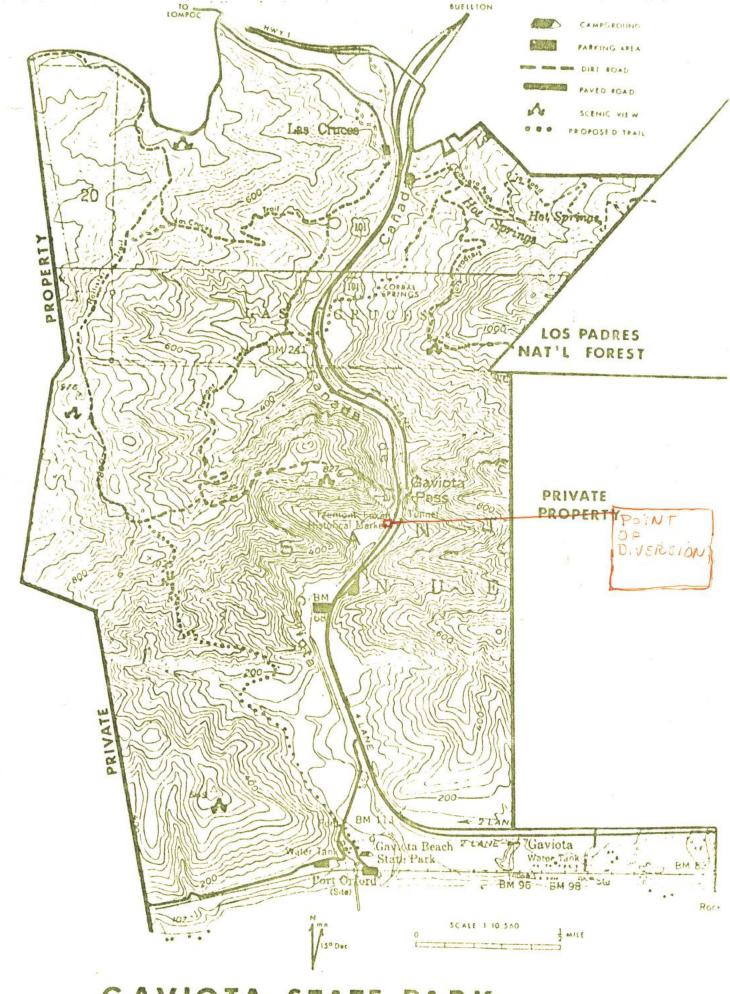
Stream	Location	100-Year Peak Flow (cfs)
Gaviota Creek	USGS Gauge No. 11120550 Below confluence with	15,571
1	West Fork Gaviota Creek Above confluence with	13,449
	West Fork Gaviota Creek Below confluence with	10,809
	East Fork Gaviota Creek	10,788
West Fork		
Gaviota Greek	Mouth San Julian Road Crossing	5,353 5,094

100-YEAR FLOOD PROFILE WEST FORK GAVIOTA CREEK

100-Year Flood Elevation Existing Conditions (feet NGVD)

Cross-Section	(feet NGVD)		
46+50	308.0		
45+00	312.2		
43+95	312.3		
43+75	312.4		
43+55	311.9		
43+00	314.6		
42+00	314.9		
41+00	316.0		
20+75	317.4		
40+50	317.4		
38+00	319.3		
37+00	320.5		
36+00	323.4		
34+00	326.8		
32+00	330.9		
30+00	333.7		
28+00	339.1		
27+00	340.9		
26+00	342.3		
23+50	346.9		
22+00	346.9		
20+00	357.2		
19+00	360.1		
17+00	365.9		
15+00	371.7		
12+50	375.4		

Note: Elevations referenced to National Geodedic Vertical Datum (NGVD).



GAVIOTA STATE PARK

mike	ROOM/STA. NO.
Low Brace	from/sta. NO.
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Returned Call Will Call Ag Information Note and Comment Re-route Investigate Return Contact Me MESSAGE/REMARKS	Reply My Signature Copy Me Forwarded Per Request
STD 7 (REV. 1-85) STD 7 (REV. 1-85) COMBINATION OF STD 7 AND 118 (RC	4796 STATE OF CALIFORNIA OUTE SLIP) ANNUAL SAVINGS \$8,300.
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